

## USE OF NANOTECHNOLOGIES IN PIG FARMING

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### Abstract

A year-long study was conducted to evaluate the use of titanium dioxide (TiO<sub>2</sub>) in pig farming. Two identical barns, each with a housing capacity of 100 pigs, slotted floors and continuous flow production were used in the study. The walls and the ceiling of one of the barns were treated with a prime sealer. The surfaces were left to dry and then a coat of paint containing TiO<sub>2</sub> nanoparticles was applied. Twenty measurements were taken during the test period to determine levels of ammonia (NH<sub>3</sub>). Average NH<sub>3</sub> emissions (converted to 100 kg live weight) were 5.04 kg in the control group and 3.92 kg in the experimental group, respectively, which corresponds to the reduction in emissions in the latter by 22.3 % (NH<sub>3</sub>). The multiple linear regression model of the relationship between NH<sub>3</sub> emissions and ambient conditions showed the greatest dependence on the weight of animals. This was corroborated by correlation statistics. The closest bond was found between NH<sub>3</sub> emissions and the weight of pigs, and it was negative ( $r$  -0.7 to -0.9).

**Key Words:** pig finishing, nanotechnologies, TiO<sub>2</sub>, ammonia, emissions

One of new elements of the modern trend is the use of nanotechnologies in various industries. Certain elements originating in nanotechnologies are also being tested in agriculture. For instance, tests are carried out using titanium dioxide (TiO<sub>2</sub>) to remove ammonia from livestock buildings. In this method, the property of metal oxides to act as catalysts in chemical processes is utilized. Chemical reactions take place on the surface of oxide molecules. Titanium dioxide is routinely used as a bleaching agent in a number of food products (E 171).

The reaction starts when TiO<sub>2</sub> is exposed to sunlight. TiO<sub>2</sub> absorbs sunlight and generates two types of carriers, electrons (e<sup>-</sup>) and holes (h<sup>+</sup>). As Fujishima et al. (1999) explained, one of important properties of TiO<sub>2</sub> is the oxidation capability of the holes, which is stronger than the reduction capability of the excited electrons. This property is utilized in industrial applications for the reduction of emissions of harmful substances into the atmosphere. In advanced research, photocatalytic properties of TiO<sub>2</sub> were utilized for sterilization and in hygiene and anti-emission applications (Li et al. (2004)). Photocatalytic oxidation can be used for the decomposition and destruction of many types of both organic and inorganic pollutants in the environment. For instance, Malloy (1999) and Watanabe et al. (1995) mentioned the use of self-cleaning floor tiles for hospitals, toilets and other places where cleanliness is of vital importance. Allen et al. (2005) demonstrated possible uses of compounds containing TiO<sub>2</sub> in city agglomerations for the reduction of nitrogen (NO<sub>x</sub>) produced by vehicles.

### Material and Methods

#### Barn description and experimental setup modifications

The animals chosen for the testing of titanium dioxide effects in livestock farming were finishing pigs. Two identical barns each with a housing capacity of 100 pigs

and continuous flow production were used. A service alley divides each barn into two halves with three pens, i.e. there is a total of 6 pens in each barn. Pigs are housed on slats. Liquid manure from storage under slotted floor is pumped to a storage tank on a weekly basis. Two exhaust fans are situated in a channel in each of the two halves of the barns. The air enters via windows along the length of one side and the door to a corridor that runs lengthwise between not only these two barns, but also a number of other barns in a row (7). Wet feed is used in all of the barns. The two barns selected are used as pig finishing facilities.

#### Experimental setup modifications

The walls and the ceiling of the experimental barn were first treated with a layer of a bonding primer. When it had dried, i.e. about a week later, the surfaces were treated with a layer of Detoxy Color paint containing TiO<sub>2</sub> nanoparticles. A paint roller was used to apply 14.5 L of the paint (= 1 L per 13.02 m<sup>2</sup>) on the pre-treated surfaces. During the experiment, the same standard type vacuum tube lighting and the same lighting schedule were used in both barns.

Because the facility was a continuous production operation, the barns housed different numbers of pigs of different weights when individual emission measurements were taken.

#### Ammonia emission measurement

Two systems, i.e. Aseko and Innova, were used to measure gas concentrations. The Innova system, which employs the photo-acoustic principle to measure concentrations of a given gas, is used as a reference system in emission measurements. The Aseko system uses electrochemical sensors. The sensors were adjusted using the Innova system. The gas monitored was ammonia (NH<sub>3</sub>). Sample collection sites with sensors were located

near underpressure fans at each of the channels in the barns monitored. Concentrations of the gas monitored were measured continuously at 7 - 10 min intervals. To determine the second component of the emission flow, ambulatory measurements of air flow were taken 4x in 24 h by the Testo 425.

### Illumination level measurements

Illumination levels were measured continuously at 10 min intervals together with some gas concentration measurements for 24 hours using the Babuc/M system and the BSR 003 probe.

### Measurement conditions

Effects of the TiO<sub>2</sub>-containing coating on the walls of the experimental barn were monitored for almost a year (30 July, 2009 – 2 July, 2010). During the year, 20 measurements were made. Both external and transient external temperatures were recorded in that period. Differences in temperature and relative air humidity in the two barns monitored were minimal and unable to influence results of gas emission measurements. Air flow in the two barns ranged from 0.09 to 0.44 m.s<sup>-1</sup>.

### Primary measured concentrations of ammonia monitored and air flow rates

The data measured in the study included concentrations of the gas monitored and air flow rates, i.e. the data indispensable for the calculation of gas emissions.

### 1. Determination of gas production

Primary data obtained by measuring concentrations of the ammonia monitored and air flow rates in two barns were used to calculate the emission flow (mg.h<sup>-1</sup>), which was then converted to the daily production of the individual gases monitored (g.pig<sup>-1</sup>.day<sup>-1</sup>). To exclude the effect of different weights of individual pigs in the two barns monitored, the number of pigs and their weight were

expressed in kg in each group. In that case, gas production per 100 kg weight of pigs over a period of 1 year (kg.100kg<sup>-1</sup>.year<sup>-1</sup>) was calculated. Basic statistics were then calculated for each group of pigs (average of all measurements, standard deviation, index compared with the reference group, difference in % and the t-test). Emission measurements were complemented with illumination level measurements.

### 2. The multiple linear regression model of the dependence of emissions of ammonia monitored on the ambient and animals

In this model, emissions of is dependent variable in both the reference and the experimental groups of pigs. The independent variables are a set of ambient parameters (temperature, relative humidity, atmospheric pressure and illumination levels) and the average live weight of pigs at the time when gas emissions were measured. The model is expressed as follows:

$$y = b_0 + ax_1 + ax_2 + ax_3 + ax_4 + ax_5 \dots r^2, \text{ where}$$

y = emissions of a gas (NH<sub>3</sub>) in both groups of pigs, as a dependent variable: kg.pig<sup>-1</sup>.day<sup>-1</sup>  
kg.100kg<sup>-1</sup>.year<sup>-1</sup>

x<sub>1</sub> = average weight of pigs at measurement (kg.pig<sup>-1</sup>).  
x<sub>2</sub> = average daily temperature at measurement (°C),  
x<sub>3</sub> = average relative air humidity (%),  
x<sub>4</sub> = atmospheric pressure at measurement (hPa)  
x<sub>5</sub> = average illumination level (lux)  
r<sup>2</sup> = coefficient of determination

### 3. Dependence of gas concentrations on ambient parameters - correlation statistics

An analysis of the relationship between the values of emissions of the gas monitored in both groups of pigs and in both conversions, and the parameters of the environment and the pigs during the monitoring period.

**Table 1. Pigs housed in barns in the period of ammonia emission measurements**

index	unit	group of pigs	
		reference	experimental
average no. of pigs	one pig	85.3	74.5
standard deviation		14.25	16.64
average weight	kg.pig <sup>-1</sup>	55.6	71.8
standard deviation		17.75	14.00
minimum no. of pigs at time of measurement	one pig	39	53
maximum no. of pigs at time of measurement	one pig	96	91
minimum average weight	kg.pig <sup>-1</sup>	33.2	47.5
maximum average weight	kg.pig <sup>-1</sup>	90.0	96.4

**Table 2. Microclimates in pig barns monitored**

parameter	groups				air pressure hPa
	reference		experimental		
	temperature	relative humidity	temperature	relative humidity	
	°C	%	°C	%	
daily average	17.41	64.9	17.16	64.6	979.96
standard deviation	4.764	9.642	5.067	9.436	6.918
daily minimum	8.0	50.6	8.1	49.4	966.1
daily maximum	22.9	83.4	23.2	81.4	988.8

**Table 3. Input data for the calculation of ammonia emissions and illumination levels**

parameter	unit	reference		experimental	
		mean	standard deviation	mean	standard deviation
air flow rate	$\text{m}^3 \cdot \text{h}^{-1}$	1875,8	476,2	1837,9	386,6
gas concentration NH <sub>3</sub>	$\text{mg} \cdot \text{m}^{-3}$	14,66	4,53	12,45	3,01
illumination level	lx	133,7	64,4	153,0	99,6

## Results and Discussion

### 1. Gas production

#### 1.1. Gas production per one pig

In view of the differences in average weights of pigs at the time of emission measurements between the two barns monitored (see tab. 1), the differences in gas production ascertained were not relevant and although neither in the production of the gas monitored was observed in the experimental group. In both barns, a continuous flow system of moving pigs into and out of the facility for slaughter was used. See Tab. 4 for an overview.

#### 1.2. Ammonia production per 100 kg live weight of pigs per year

Because weight of individual pigs in the barns was different, the numbers of pigs in each of the barns were expressed in pig units of 100 kg live weight as a common denominator for the purpose of gas emission calculations. In this case, differences in gas emissions were obvious and statistically significant. NH<sub>3</sub> emissions dropped by 21.3 % ( $P < 0.001$ ). See Tab. 5 for an overview. For a comparison with the farrowing facility, the walls and the ceiling of which were treated with a paint containing TiO<sub>2</sub> nanoparticles. Sows were moved in 3 – 5 days prior to farrowing and moved out 21 days post farrowing, including about 10 piglets per sow, ammonia emissions were reduced by 30.4 % and nitrogen monoxide emissions by 3.9 % (Guarino et al. (2007)).

### 2. The linear regression model of the dependence of emissions of gas monitored on the ambient and animals

#### 2.1. Conversion per pig

The multiple linear dependence model, where monitored ammonia emissions was the dependent variable and parameters of the environment and animals the independent variables, showed a relatively high degree of dependence ( $r^2 > 0.82$ ). The calculated model equations are suitable for NH<sub>3</sub> emission predictions ( $F < 0.05$ ). The statistical significance of live weight linear coefficients of the pigs ( $x_1$ ) was very high in both groups of pigs ( $P < 0.001$ ) for NH<sub>3</sub> emissions. In the NH<sub>3</sub> emissions model of the reference group, the atmospheric pressure  $x_4$  was also important ( $P < 0.05$ ). Effects of the other variables were not statistically significant.

#### 2.2. Conversion to 100 kg live weight of pigs

It describes the dependence model for NH<sub>3</sub> emissions. The coefficients of determination ( $r^2$ ) of the reference and the experimental groups were even higher than 0.98 and 0.89, respectively. The calculated equations are suitable for the predictions of NH<sub>3</sub> emission ( $F < 0.001$ ) in the reference and ( $F < 0.01$ ) in the experimental groups of pigs. As in the per one pig conversion, weight of the animals ( $x_1$ ) plays an important role ( $P < 0.001$ ) but has a contrary effect on emission levels, i.e. when this conversion is made, young animals produce the largest amounts of NH<sub>3</sub>. In the reference group, relative air

humidity ( $x_3$ ) and atmospheric pressure ( $x_4$ ) were also statistically significant ( $P < 0.05$  and  $P > 0.01$ , respectively). In addition to the already-mentioned animal weight ( $x_1$ ), parameters that also played a role in the experimental group were air temperature ( $x_2$ ,  $P < 0.05$ ) and illumination levels ( $x_5$ ,  $P < 0.05$ ). These relationships are presented in Tab. 6.

### 3. Dependence of ammonia concentrations on ambient parameters and animals- correlation

#### 3.1. Emissions per one pig

By far the highest correlation was found between  $\text{NH}_3$  emissions and animal weight. The correlation between

these elements (with the exception of  $\text{NH}_3$  in the experimental group) showed a very close relationship. A relatively close relationship was also found in  $\text{NH}_3$  emissions in the experimental group ( $r = 0.81$ ). Only a loose relationship was found between other ambient parameters and emission of the gas monitored.

#### 3.2. Conversion of emissions per 100 kg weight

The closest relationship between gas emissions calculated per 100 kg of pig weight and the weight of the pigs was found in  $\text{NH}_3$ : the relationship was very close ( $r = -0.97$ ) in the reference group and close ( $r = -0.76$ ) in the experimental group. The relationships are summarized in Table 7.

**Table 4. Ammonia production per one pig per year ( $\text{g.pig}^{-1}.\text{day}^{-1}$ )**

n	groups				index	$\Delta$ (%)
	reference		experimental			
	mean	standard deviation	mean	standard deviation		
20	7.540	0.943	7.540	1.394	1,000	0,0

**Table 5. Ammonia production per 100 kg of pig weight per year ( $\text{kg.100}^{-1}.\text{year}^{-1}$ )**

n	groups				index	$\Delta$ (%)	statistical significance $P <$
	reference		experimental				
	mean	standard deviation	mean	standard deviation			
20	5.037	1.104	3.916	0.570	0.777	-22.3	0.001

**Table 6. The linear model of the dependence of emissions of ammonia monitored on ambient conditions and animals**

	group	independent variable					$b_0$	$r^2$	F
		$x_1$	$x_2$	$x_3$	$x_4$	$x_5$			
<b>per one pig conversion</b>									
$\text{NH}_3$	ref.	0.041 <sup>***</sup>	-0.030	-0.032	-0.066 <sup>*</sup>	0.001	73.04	0.873	0.05
	exp.	0.015	0.154	0.003	0.036	-0.008	-30.78	0.840	0.05
<b>per 100 kg weight conversion</b>									
$\text{NH}_3$	ref.	-0.055 <sup>***</sup>	-0.008	-0.044 <sup>*</sup>	-0.036 <sup>**</sup>	-0.001	47.01	0.981	0,001
	exp.	-0.051 <sup>***</sup>	-0.089 <sup>*</sup>	-0.003	0.019	-0.006 <sup>*</sup>	-12.58	0.892	0,01

\* $P < 0.05$ , \*\* $P < 0.01$ , \*\*\* $P < 0.001$ , NS = statistically not significant

**Table 7. Correlation dependence between NH<sub>3</sub> emissions and ambient parameters**

group	ambient parameters				
	pig weight	temp.	relative humidity	air pressure	illumination level
	kg.pig <sup>-1</sup>	°C	%	hPa	lux
g.pig <sup>-1</sup> .day <sup>-1</sup>					
reference	0.848	-0.283	0.197	-0.344	-0.149
experimental	0.604	0.811	-0.612	0.417	0.252
kg.100kg <sup>-1</sup> .year <sup>-1</sup>					
reference	-0.970	0.262	-0.263	-0.037	0.173
experimental	-0.761	-0.110	0.150	0.137	-0.256

## Conclusion

A year-long monitoring of the effects of TiO<sub>2</sub> nanoparticles demonstrated the functionality of this approach to the reduction of ammonia concentrations. Ammonia emissions were reduced on average by 22.3 % , (P< 0.001). It is not yet quite clear, how long the TiO<sub>2</sub> coating will retain its functionality on the wall plastering. Monitoring of the nanoparticle coating will continue and, moreover, light of different wavelengths will be used for its illumination with the objective to improve the efficiency of ammonia and greenhouse gases elimination.

## References

- Allen,N.S., Edge,M., Sandoval,G., Verran,J., Stratton,J., Maltby,J. (2005): Photo catalytic coatings for environmental applications. *Photochem.Photobiol.* 81 (2):279-290
- Fujishima,A., Hashimoto,K., Watanabe,T.(1999): TiO<sub>2</sub> photo catalysis: fundamentals and applications. BKC,Inc., Tokyo.
- Guriano,M., Costa, A., Porro,M.(2007): Photocatalytic TiO<sub>2</sub> coating to reduce ammonia and greenhouse gases concentration and emission from animal husbandries. *Bioresour.Technol.*,(2007),doi:10.1016/j.biortech.2007.04.025.
- Li,F.B., Li,X.Z., Ao, C.H., Hou, M.F., Lee, S.C.(2004): Photo catalytic conversion of NO using TiO<sub>2</sub> – NH<sub>3</sub> catalysts in ambient air environment. *Appl. Catal. B: Environ.*54:275-283.
- Malloy,H.(1999): Environmentally friendly ceramic tile. *Ceram.Ind.*149. (10):37-42.
- Watanabe,T., Kojima,E., Norimoto,K, Saeki,Y.(1995): Fabrication of TiO<sub>2</sub> photo catalytic tile and practical applications. *Fourth Euro Ceramics*, 11:175-180.

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